



Research article

Exploring the potential of underrated yet versatile crop *Lunaria annua* L.: New insights into honesty plant

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ABSTRACT

Lunaria annua is an underexplored crop with promising potential for use in various industries: food, oil etc. It grows rapidly, has low maintenance and high resistance to pathogens, which makes it a promising crop for sustainable cultivation. Herein, we investigated nutritional value, phenolic profile, and bioactive properties of *L. annua* seeds collected in the wild. The results we obtained indicate that seeds of *L. annua* have high energy value (406.6 kcal/100 g dw), with high share of carbohydrates (65.2 g/100 g dw), followed by proteins (26.47 g/100 g dw) and fat (4.46 g/100 g dw). Free sugar analysis showed presence of fructose, glucose and sucrose (0.77 g, 0.189 g and 3.85 g/100 g dw, respectively). Oxalic and malic acid were quantified <1.0 g/100 g dw and fumaric acid only in trace. Monounsaturated fatty acids were predominant over polyunsaturated and saturated fatty acids (erucic, nervonic and oleic acid were dominant). The content of flavonoid compounds was several times higher than the content of phenolic acids (58.5 vs 9.4 mg/g). Methanolic extract showed better antioxidant potential than dichloromethane extract (EC₅₀ = 0.32 vs. 1.25 mg/mL, in TBARS test). The antibacterial activity ranged from 0.50 to 2.00 mg/mL, with *E. coli* showing the highest susceptibility to the methanolic extract (MIC: 0.25 mg/mL, MBC: 0.50 mg/mL). The other tested microorganisms showed quite uniform susceptibility to the extracts, but better than the positive controls E211 and E224. The tested samples demonstrated encouraging antifungal/anticandidal properties, and no cytotoxic effects onto spontaneously immortalized keratinocytes cell line HaCaT, indicating its safe application in terms of dermal exposure. Overall, our results indicate that *L. annua* seeds may be considered as source of compounds used in food and pharmaceutical industry and a candidate for functional ingredient that provides additional health benefits beyond basic nutritional requirements.

1. Introduction

Many plants belonging to the Brassicaceae family are highly valued crops widely consumed across the globe, not only for their

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distinctive flavors but also for their health beneficial properties, which are associated to their phytochemical composition. Among these compounds, phenolics, isothiocyanates, carotenoids, minerals, and vitamins are most often recognized as bioactive constituents, used to enhance daily requirements for nutrients and biologically active compounds [1]. Given the rising demand for functional food ingredients that provide additional health benefits, there is an increasing interest within the underexplored species of the Brassicaceae family and their potential for larger-scale cultivation and utilization.

Lunaria annua L. (aka honesty or money plant) is one of the most underrated members of Brassicaceae family, even though it offers great potential for industrial-scale cultivation. It grows rapidly widespread across Asia and Europe, thrives with minimal maintenance, and is quite resistant to pathogenic microorganisms [2]. The ability of money plant to self-seed further reduces labor costs, thus enhancing profitability upon cultivation. Given its low maintenance requirements, high abundance of oils in seeds and their versatile application in the food and oil industries, *L. annua* can be considered a sustainable, underexplored crop that could be efficiently cultivated on abandoned land [3]. Estimates suggest that agricultural abandonment led to creation of 120 million hectares of unused land, especially in remote locations in Europe [4], which presents a significant opportunity for *L. annua* cultivation given the fact it does not require farmland of the highest quality and offers revenue without directly competing with other crops in the market. Through repurposing of this farmland, we can create opportunities for cultivation of this underutilized species in unfavorable conditions.

This seems to be of particular interest, given that research data on chemical profile and bioactive properties of *L. annua* are quite scarce. To this date, chemical analyses have shown that seeds contain 30–35 % oil, of which 67 % consists of long-chain fatty acids. The fatty acid profile comprises 44 % erucic acid (C22:1) and 23 % nervonic acid (C24:1), both of which offer various potential applications in the pharmaceutical and oil industries [3]. In addition, the aerial parts of *L. annua* were found to contain 12 glucosinolates, with glucohesperin, glucoalyssin, glucobrassicinapin, glucoputranjivin, and hex-5-enyl glucosinolate being the most abundant, along with anthocyanins [5,6]. As for the bioactive properties, Katanić Stanković et al. (2022) demonstrated anti-inflammatory and antioxidant activities of *L. annua* herba [2], while Blazevic et al. (2013; 2019) showed cytotoxic potential against lung and breast tumor cell lines [5,6].

Even though seeds are mostly used as a food ingredient, detailed analyses of their nutritional value, chemical profile and bioactive properties are still lacking. Knowing this may reveal whether honesty plant can be considered a sustainable energy source in the diet or a candidate for functional food. This study aimed to chemically characterize the seeds of *L. annua*, with a focus on their nutritional and energy value, as well as their content of free sugars, organic acids, and fatty acids. We also analyzed phenolic profile, given the well-known bioactive properties of these compounds [7]. Furthermore, we assessed antibacterial, antifungal, antioxidant, and cytotoxic properties of *L. annua* as to explore its pharmaceutical potential to provide additional health benefits beyond basic nutritional requirements. This selected metabolite analysis and exploration of honesty plant biological activities could possibly allow prioritization of its cultivation and end-user application in food and pharmaceutical industry, supporting sustainable and economically feasible principles of bio-based industry as well agriculture.

2. Material and methods

2.1. Collection of plant material

Aerial parts of *L. annua* plants were collected in Mionica, Serbia, in August 2021, when pods (siliques containing seeds) were dried. Authentication was carried out by dr Dejan Stojković at the Institute for Biological Research “Siniša Stanković,” National Institute of the Republic of Serbia, University of Belgrade. A voucher specimen, labeled La-IBRSS-2021, has been deposited in the Plant Collection Unit of the Mycological Laboratory at IBISS. Following collection, the pods were cleaned of the central membrane and debris, ground into a fine powder, and stored at 4 °C for further analysis.

2.2. Preparation of methanolic and dichloromethane extracts

Powdered *L. annua* seeds (10 g) were extracted by stirring with 250 mL of methanol or dichloromethane at –20 °C overnight, following the method of Vaz et al. [8]. The extraction solvents were selected to investigate potential bioactive compounds across solvents with different polarities. The extract was sonicated for 15 min, centrifuged at 4000 g for 10 min, and filtered through Whatman No. 4 paper. The residue was re-extracted twice with 150 mL portions of methanol or dichloromethane. The combined extracts were evaporated to dryness at 40 °C using a Büchi R-210 rotary evaporator and re-dissolved in 30 % ethanol.

2.3. Standards and reagents

Acetonitrile (99.9 %), n-hexane (95 %), and ethyl acetate (99.8 %) of HPLC grade were sourced from Fisher Scientific (Lisbon, Portugal). Methanol and other analytical-grade chemicals were obtained from Merck (Darmstadt, Germany). FAME reference standard mixture (47885-U) and various individual standards for fatty acids, tocopherols, sugars, organic acids, and phenolic compounds were purchased from Sigma Chemical Co. (St. Louis, MO, USA). Additional reagents and media, including Mueller-Hinton agar (MH), malt agar (MA), and cell culture components, were acquired from Torlak Institute (Belgrade, Serbia) and HyClone (Logan, USA). Water was purified using a Milli-Q system (TGI Pure Water Systems, USA).

2.4. Chemical characterization of *L. annua*

2.4.1. Nutritional value

The seed powder was analyzed for its protein, fat, carbohydrate, and ash content following the procedures outlined by the Association of Official Analytical Chemists [9]. Crude protein content (conversion factor: 6.25) was determined using the macro-Kjeldahl method, which involves digestion, distillation, and titration. Crude fat content was measured by extracting the powdered sample with petroleum ether using a Soxhlet apparatus. Ash content was determined by incinerating the sample at 500 ± 15 °C for 5 h in a muffle furnace. Total carbohydrates were calculated by difference, and energy was estimated using the following equation.

$$\text{Energy (kcal)} = 4 \times (\text{g}_{\text{protein}} + \text{g}_{\text{carbohydrate}}) + 9 \times (\text{g}_{\text{fat}}).$$

2.4.2. Sugars composition

To extract free sugars, mixture of ethanol:water (80:20, 40 mL) solution and Internal Standard (IS, melezitose, 25 mg/mL, 1 mL) was added to the powdered seeds (1 g). Using a water bath, extraction was performed for 1 h and 30 min at 80 °C, as described previously by Spréa et al. [10]. Subsequently, ethanol was evaporated, and the obtained aqueous portion was washed successively with 10 mL of ethyl ether (3 x). After concentration at 40 °C, the residue was re-dissolved in distilled water to a final volume of 5 mL and filtered through 0.2 µm nylon filters into vials. The analysis was conducted using a High-Performance Liquid Chromatography (HPLC) system, which included a pump (Knauer Smartline 1000), a degasser (Smartline Manager 5000), and an auto-sampler (AS-2057 Jasco), coupled with a refractive index detector (Knauer Smartline 2300 RI detector), operating under the same conditions as previously described by the authors [10]. Sugar identification was performed by comparing the relative retention times of sample peaks with those of standards. The resulting data were analyzed using Clarity 2.4 Software (DataApex 1.7 Prague, Czech Republic). Quantification was carried out based on the RI signal response of each standard, using the internal standard (IS, melezitose, 25 mg/mL) method and calibration curves derived from commercial standards of each compound. The results were expressed in grams per 100 g of dry weight (g/100g dw).

2.4.3. Organic acids composition

To evaluate organic acids, 1 g of powdered *L. annua* seeds was extracted with 25 mL of metaphosphoric acid for 25 min with magnetic stirring at room temperature, then filtered through Whatman No. 4 paper, following the procedure of Barros et al. [11]. The solutions were filtered through 0.2 µm nylon filters into vials and analyzed using an Ultra-Fast Liquid Chromatography system with a reverse phase C18 SphereClone column (Phenomenex, 5 µm, 250 × 4.6 mm), maintained at 35 °C, and coupled to a diode array detector (UFLC-DAD; Shimadzu Cooperation, Kyoto, Japan). Quantification of organic acids was performed by comparison of the area of their peaks (at 215 nm) with calibration curves obtained from commercial standards of each compound, including: oxalic, quinic, malic, ascorbic, citric and fumaric acids. All were purchased from SigmaAldrich, St. Louis, USA. Obtained results were analyzed using the LabSolutions Multi LCPDA (Shimadzu Corporation, Kyoto, Japan), and expressed as g per 100 g of dry weight.

2.4.4. Fatty acids composition

Fatty Acid Methyl Esters (FAMES) were determined from the lipid fraction obtained through Soxhlet extraction. The lipid portion of the sample was derivatized via a transesterification reaction, following the method of Obodai et al. [12]. The procedure consisted of following: addition of a methanol:sulfuric acid:toluene solution (2:1:1, v/v/v, 5 mL) to the lipid fraction previously extracted and incubating for 12 h at 50 °C. After incubation, 3 mL of distilled water and 3 mL of diethyl ether were added to the sample and mixed thoroughly using a vortex. The organic phase (which contained FAMES) was removed, dehydrated with anhydrous sodium sulfate and filtered with nylon filters (0.2 µm) for chromatographic analysis. Profile of fatty acids was determined using Gas Chromatography (GC) with Flame Ionization Detection (FID), and YOUNG IN Chromass 6500 GC System instrument containing a *split/splitless* injector. Identification and quantification of fatty acids was made after comparing relative retention times of FAME peaks obtained from samples with those of standards (standard mixture 47885-U, Sigma, St. Louis, MO, USA). Obtained results were recorded and processed using Clarity 4.0.1.7 Software (DataApex, Prague, Czech Republic). They were further expressed as the relative percentage (%) of each fatty acid.

2.4.5. Phenolic compounds composition

Profiles of the phenolic acids in extracts were determined using High Performance Liquid Chromatography (HPLC) on a Dionex Ultimate 3000 UPLC (Thermo Fisher Scientific, San Jose, CA, USA) equipped with a Diode Array Detector (DAD), as was previously published [13]. For the separation, we used a Waters Spherisorb S3 ODS-2 C18 column (3 µm, 150 × 4.6 mm, Waters Milford, MA, USA), operating at 35 °C. For the mobile phase 0.1 % formic acid (A) and acetonitrile (B) were used. The elution gradient was 15 % B (5 min), 15–20 % B (5 min), 20–25 % B (10 min), 25–35 % B (10 min), and 35–50 % B (10 min). Mass detection was carried out using a LTQ Orbitrap XL mass spectrometer (Thermo Scientific, San Jose, CA, USA), with an ESI electrospray ionization source, with system operating at 325 °C, and a spray voltage of 5 kV with a capillary voltage of –20 V. The spectra were recorded in negative ion mode between 100 and 1500 *m/z*. Obtained results were further analyzed with Xcalibur® program (version 2.2.0.48; Thermo Fisher Scientific, San Jose, California, USA). The phenolic compounds in the samples were characterized by comparing their UV–vis spectra, mass spectra, and retention times with those of standards and literature sources, when available. For quantitative analysis, a

calibration curve was created by injecting known concentrations of various standards, and the compounds were quantified based on the UV–Vis signal of the commercial standards at their λ maximum. When standards were unavailable, other compounds with the same phenolic group were used. The results were expressed in mg per g of extract.

2.5. Antimicrobial activity

2.5.1. Antibacterial activity

Antibacterial activity assay was performed using microdilution method described by Soković et al. [14]. We tested following Gram (+) bacteria: *Staphylococcus aureus* (ATCC 11632), *Bacillus cereus* (food isolate) and *Listeria monocytogenes* (NCTC 7973) as well as Gram (–) bacteria *Escherichia coli* (ATCC 25922), *Enterobacter cloacae* (ATCC 35030) and *Salmonella* Typhimurium (ATCC 13311). Bacterial strains were cultured overnight at 37 °C in Tryptic Soy Broth and then adjusted to a concentration of 1.0×10^5 CFU/mL using sterile saline. Samples dissolved in a 30 % ethanol solution were added to Tryptic Soy Broth (TSB, 100 μ L) medium with 1.0×10^4 CFU of bacterial inoculum per well. After incubation at 37 °C, 24 h, *p*-iodonitrotetrazolium chloride (0.2 mg/mL, 40 μ L) was added to each well, and the plate was further incubated for 60 min at 37 °C to allow color development. The lowest concentration showing a noticeable reduction in color intensity (light red compared to the intense red in the control well, which had no extract) or no color at all, was defined as the minimal inhibitory concentration (MIC). The minimal bactericidal concentration (MBC) was determined by serial sub-cultivation of 2 μ L into wells with 100 μ L of broth, followed by a further 24-h incubation at 37 °C. The lowest concentration with no visible growth was identified as the MBC, indicating a 99.5 % kill of the original inoculum. Commercial preservatives E211 and E224 served as positive controls.

2.5.2. Antifungal activity

Antifungal assay was performed using a protocol previously described in a study by Petrović et al. (2022) [15]. For antifungal activity assay we used: *Aspergillus fumigatus* (ATCC 9197), *Aspergillus niger* (ATCC 6275), *Aspergillus versicolor* (ATCC 11730), *Penicillium funiculosum* (ATCC 36839), *Penicillium verrucosum* var. *cyclopium* (food isolate), *Trichoderma viride* (IAM 5061). The microorganisms are deposited at Mycological laboratory, Department of Plant Physiology, Institute for Biological research “Siniša Stanković”, National Institute of Republic of the Serbia, University of Belgrade, Serbia. Before performing microdilution method, fungal spores were washed off the surface of agar plates using sterile 0.85 % saline containing 0.1 % Tween 80 (v/v). Samples dissolved in a 30 % ethanol solution were added to Malt broth medium, which was followed by the addition of the fungal inoculum, and further 5-days incubation at 25 °C. The lowest concentrations showing a significant reduction in mycelial growth (observed under a binocular microscope) were defined as MICs. The minimal fungicidal concentrations (MFC) were determined by serial sub-cultivation of the tested sample dissolved in medium (sample volume 2 μ L), followed by further incubation for 72 h at 25 °C. The lowest concentration with no visible growth was identified as the MFC, indicating 99.5 % killing of the original inoculum. Commercial preservatives E211 and E224 were used as positive controls.

2.5.3. Anticandidal activity

For anticandidal assay, following isolates were used: *Candida albicans* 475/15, *C. albicans* 13/15, *C. albicans* 17/15, *C. parapsilosis* ATCC 22019, *C. tropicalis* ATCC 750 and *C. krusei* H1/16. Clinical isolates were obtained from the ENT Clinic, Clinical Hospital Centre Zvezdara, Belgrade, Serbia. Strains of *Candida* spp. were identified on CHROMagar plates (Biomérieux, France) and maintained on Sabouraud Dextrose Agar (Merck, Germany). Minimum inhibitory concentrations (MIC) and minimum fungicidal concentrations (MFC) were determined using a microdilution method with modifications described by Smiljkovic et al., 2018 [16]. Microplates, containing serially diluted samples, broth, and fungal inoculum, were incubated at 37 °C for 24 h. The MIC values were determined after incubation as the lowest concentrations showing no visible growth under a microscope. MFC values were determined by serial sub-cultivation of broth (10 μ L) with no visible fungal growth into wells containing fresh broth (100 μ L), followed by overnight incubation at 37 °C. Ketoconazole (1 μ g/mL), a commercial antifungal drug, was used as a positive control.

2.6. Antioxidant activity

2.6.1. TBARS

Inhibition of Thiobarbituric Acid Reactive Substances (TBARS) was performed following the procedure described by Reis et al. (2012) [17]. Porcine (*Sus scrofa*) brains, obtained from slaughterhouse, were dissected and homogenized using a Polytron in an ice-cold Tris–HCl buffer (20 mM, pH 7.4) to create a 1:2 w/v brain tissue homogenate. The sample was subsequently centrifuged at 3000 g for 10 min and the aliquot of supernatant (100 μ L) was incubated with different concentrations of sample solutions (200 μ L), FeSO₄ (10 mM, 100 μ L) and ascorbic acid (0.1 mM, 100 μ L) at 37 °C for an hour. After stopping the reaction with trichloroacetic acid (28 % w/v, 500 μ L) the solution of thiobarbituric acid (2 % w/v, 380 μ L) was added and the mixture was heated for 20 min at 80 °C. Precipitated protein was removed after centrifuging the sample at 3000 g for 10 min, and the color intensity of the malondialdehyde (MDA)–TBA complex in the supernatant was measured at 532 nm. The inhibition ratio (%) was calculated according to the following: inhibition ratio (%) = $[(A - B)/A] \times 100$ %, where A and B are the absorbance of the control (mixture without extract) and the sample solution, respectively.

Table 1
Nutritional, energetic value and hydrophilic compounds of *Lunaria annua* seeds (mean \pm SD, $n = 3$).

Nutritional value	(g/100g dw)	Student <i>t</i> -Test
Fat	4.46 \pm 0.06	<0.001
Proteins	26.47 \pm 0.04	<0.001
Ash	3.9 \pm 0.1	<0.001
Carbohydrates	65.2 \pm 0.1	<0.001
Energy (Kcal/100g dw)	406.6 \pm 0.1	<0.001
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Free sugars	(g/100g dw)	
Fructose	0.77 \pm 0.07	<0.001
Glucose	0.189 \pm 0.001	<0.001
Sucrose	3.85 \pm 0.03	<0.001
Total	4.8 \pm 0.1	<0.001
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Organic acids	(g/100g dw)	
Oxalic acid	0.23 \pm 0.01	<0.001
Malic acid	0.48 \pm 0.03	<0.001
Fumaric acid	trace	<0.001
Total	0.70 \pm 0.03	<0.001

Mean statistical differences obtained by *t*-Student test.

2.7. Cytotoxicity towards HaCaT cell line

Crystal violet test was applied to evaluate cytotoxicity towards spontaneously immortalized keratinocytes (HaCaT cell line) as described in a study by Petrović et al. (2021) [18]. HaCaT cells were cultured in an incubator with 5 % CO₂ at 37 °C using high-glucose Dulbecco's Modified Eagle Medium (DMEM), which was supplemented with fetal bovine serum (10 %, FBS), 2 mM l-glutamine, and antibiotic-antimycotic (1 %, Invitrogen). Cells (10⁴) were seeded in a 96-well microtiter plate with an adhesive bottom. After 48 h, the media was removed, and the cells were treated overnight (in triplicate) with varying concentrations of the tested samples. The cells were further washed twice with PBS and then stained with a crystal violet solution (0.4 %) for 20 min. The stain was then removed and the cells were rinsed with tap water and allowed to air dry at room temperature. Methanol was then added to dissolve the bound crystal violet residues. The resulting absorbance was measured at 570 nm using a plate reader (Multiskan™ FC Microplate Photometer, Thermo Scientific). Potassium dichromate (K₂Cr₂O₇) was used as a positive control and PBS as a negative one. The results were expressed as the IC₅₀ (%) value in µg/mL, indicating the concentration required for 50 % cell viability in comparison to the untreated control. The cytotoxic activity of the extracts on the HaCaT cell line was categorized using the following criteria: IC₅₀ ≤ 20 µg/mL = highly cytotoxic, IC₅₀ ranging between 21 and 200 µg/mL = moderately cytotoxic, IC₅₀ ranging between 201 and 400 µg/mL = weakly cytotoxic, and IC₅₀ > 401 µg/mL = no cytotoxicity [19].

2.8. Statistical analysis

All experiments were performed in triplicate, and the results are presented as mean values with the corresponding standard deviations (SD). The data were analyzed using a Student's *t*-test to assess the significance of differences between two samples at a 5 % significance level (Version 25, IBM Corp., New York, NY, USA), except for antimicrobial activity.

3. Results and discussion

Until recently, *L. annua* has primarily been cultivated for its distinctive silique seed pods, commonly used in dried floral arrangements and for home decoration. However, recent studies revealed that this low-maintenance plant also possesses an unusual chemical profile, which has potential applications in various industries. This data offer promising new uses for honesty plant that go far beyond its decorative use. Thus, herein we present a comprehensive nutritional characterization of *L. annua* seeds, including analyses of free sugars, fatty acids, and a detailed profile of phenolic compounds, along with their associated bioactive properties.

Results regarding selected hydrophilic compounds of *L. annua* samples are presented in Table 1. They indicate that seeds are valuable source of carbohydrates (65.2 g/100 g dw) followed by proteins (26.47 g/100 g dw) and fat (4.46 g/100 g dw). This is quite important given that seeds are used as a food ingredient, but so far, their nutritional profile has been poorly investigated. Our analysis indicates that seeds are valuable source of macronutrients, which along with other beneficial chemical constituents make them good candidate for functional food. Due to their high carbohydrate content, seeds can serve as an excellent primary source of energy in the diet. Furthermore, since rich source of protein, they are particularly valuable in vegetarian and vegan diets that depend exclusively on plant-based protein sources. Total fat content in the sample is generally considered low, but the high abundance of beneficial fatty acids makes the seeds desirable in specialized diets with limited lipid intake.

Regarding free sugar analysis, disaccharide sucrose was present in highest amount 3.85 g/100 g dw. While this amount is not particularly high, one should be cautious upon consumption of *L. annua*, as elevated sucrose intake has been linked with a higher risk of developing type 2 diabetes and elevated plasma triglyceride levels [20]. As for the organic acid profile, our results revealed low

Table 2
Fatty acids profile of the *Lunaria annua* seeds (mean \pm SD, $n = 3$).

Fatty acids	Relative percentage (%)	Student <i>t</i> -Test
C14:0	0.124 \pm 0.003	<0.001
C15:0	0.109 \pm 0.002	<0.001
C16:0	3.12 \pm 0.01	<0.001
C16:1	0.220 \pm 0.001	<0.001
C18:0	0.79 \pm 0.01	<0.001
C18:1n9c	21.12 \pm 0.04	<0.001
C18:2n6c	6.3 \pm 0.09	<0.001
C18:3n3	0.721 \pm 0.06	<0.001
C20:1	1.6 \pm 0.1	<0.001
C20:5n3	0.108 \pm 0.001	<0.001
C22:0	0.434 \pm 0.001	<0.001
C22:1n9	38.10 \pm 0.07	<0.001
C22:2	0.183 \pm 0.001	<0.001
C24:0	0.393 \pm 0.001	<0.001
C24:1	26.73 \pm 0.05	<0.001
SFA	4.97 \pm 0.02	<0.001
MUFA	87.76 \pm 0.05	<0.001
PUFA	7.27 \pm 0.03	<0.001

C14:0 - Myristic acid, C15:0 - Pentadecanoic acid, C16:0 - Palmitic acid, C16:1 - Palmitoleic acid, C18:0 - Stearic acid, C18:1n9c - Oleic acid, C18:2n6c - Linoleic acid, C18:3n3 - α -Linolenic acid, C20:1 - Eicosenoic acid, C20:5n3 - Eicosapentaenoic acid (EPA), C22:0 - Behenic acid, C22:1n9 - Erucic acid, C22:2 - Docosadienoic acid, C24:0 - Lignoceric acid, C24:1 - Nervonic acid, SFA - saturated fatty acids, MUFA - monounsaturated fatty acids, PUFA - polyunsaturated fatty acids. Mean statistical differences obtained by *t*-Student test.

share of oxalic and malic acid (<1.00 g/100 g dw), with traces of fumaric acid. Even though organic acids have been connected to various health benefits [21], their low concentration in *L. annua* does not diminish the potential health benefits provided by other bioactive compounds present in the plant. The results of fatty acid analysis are presented in Table 2. They show a higher prevalence of monounsaturated fatty acids compared to polyunsaturated and saturated fatty acids, with erucic (C22:1n9), nervonic (C24:1) and oleic (C18:1n9c) being the most prevalent fatty acids found in the seeds of the honesty plant: 38.10 %, 26.73 % and 21.12 %, respectively. Their high content is significant, as plant-based foods are preferred sources of MUFAs for preventing coronary heart disease [22,23]. Previously published data indicated that honesty plant has been explored as source of oils used as lubricants in industry, due to relatively high oil yield from seeds. From 1000 to 2000 kg/ha, oil yield from seeds was ~30–35 %, with nervonic (C24:1) and erucic acid (C22:1) as the main constituents (23 % and 44 % respectively) [23,24]. Results on MUFA abundance align with our results; however, additional research is necessary to fully elucidate potential of these fatty acids in pharmaceutical industry. So far, research showed that high abundance of nervonic acid in the *L. annua* seeds has re-myelinating properties, which may be significant in treating neurological disorders accompanied with loss of myelin [25]. As for the erucic acid, even though data suggest it has adverse effects in experimental animals [26], they lack confirmation in humans. Furthermore, people who consume food rich in this compound had no signs of intoxication. Thus, the use of *L. annua* seeds in cuisine as a substitute for mustard consistently showed no toxic effects on humans.

Results of the phenolic compounds profile by LC-DAD-ESI-MSⁿ are presented in Table 3. The phenolic compounds were detected in methanolic extract (Figs. 1 and 2), whereas dichloromethane extract showed no traces of phenolic compounds. The variation in profiles of phenolic compounds in the examined extracts can be attributed to the differing polarities of the solvents used in the extraction. Organic solvents with higher polarity, such as methanol, are more effective at extracting more polar compounds, resulting in the extraction of higher content of phenolic compounds (polar compounds) than dichloromethane, a solvent with lower polarity [27]. Results presented in Table 3 show the compounds identified in sample based on their chromatographic behavior, UV-Vis data, and comparison of fragmentation patterns with those previously reported in the literature. The MS² spectral data for each compound are presented in the Supplementary Material of this article.

Twenty-six compounds were determined in the methanolic extract. Flavonoids were the compounds found in the greatest amount (total of 58.5 mg/g of extract). The total phenolic acids content was 9.4 mg/g of extract, with vanillic acid rutinoside and a vanillic acid derivative being the most representative compounds (5.71 mg/g of extract each, coeluted in the same peak). **Peak 1** was assigned as taxifolin hexoside ($\lambda_{\max} = 285$ nm; $[M - H]^-$ m/z 465) due to the neutral loss of 162 Da (indicating the loss of a hexose moiety) that yielded m/z 303 as the aglycone ($[M-H-162]^-$), and the characteristic fragments at m/z 285 ($[M-H-162-18]^-$), m/z 177 and m/z 125, both corresponding to the ring fission of taxifolin [28–30]. Taxifolin (**peak 16**; $\lambda_{\max} = 300$ nm; $[M - H]^-$ at m/z 303) was confirmed using authentic standards.

Thirteen compounds were tentatively identified as quercetin derivatives. Among them, quercetin-3-*O*-rutinoside (Rutin) (**peak 13**; $\lambda_{\max} = 259/351$ nm; $[M - H]^-$ at m/z 609), and quercetin-3-*O*-glucoside (**peak 15**; $\lambda_{\max} = 274/319$ nm; $[M - H]^-$ at m/z 463), were identified by comparison with an authentic standard. Quercetin-*O*-glycosides were characterized by the presence of the aglycone at m/z 301 as main fragment ion, which was attributed to quercetin [31,32]. In addition, they had fragmentation patterns that indicated the presence of sugar moieties linked to the aglycone due to the neutral losses of 146 Da, 162 Da, 324 Da (162 Da +162 Da), 308 Da (162

Table 3
Phenolic composition of *L. annua* seeds oil extracts (mean \pm SD).

Peak n. ^o	Rt (min)	λ_{\max} (nm)	[M – H] (m/z)	MS ² fragments (m/z)	Tentative identification	<i>L. annua</i> (MeOH) (mg/g of extract)	Student <i>t</i> -Test
1	4.97	285	465	MS ² [465]: 285 (100), 125 (33), 177 (14), 303 (14)	Taxifolin hexoside [28–31]	2.73 \pm 0.08	<0.001
2	5.15	267/ 327	787	MS ² [369]: 463 (100), 301 (98), 179 (44), 151 (37), 625 (3)	Quercetin trihexoside [63,64]	1.4 \pm 0.07	<0.001
3	5.43	271/ 312	1095	MS ² [1095]: 463 (100), 301 (58), 609 (31), 771 (18), 625 (13), 179 (1)	Quercetin tetrahexoside-rhamnoside [36]	0.85 \pm 0.02	<0.001
4a ^a	5.71	272/ 319	369	MS ² [369]: 249 (100), 189 (80), 145 (76), 369 (75), 207 (72), 163 (56), 119 (50), 223 (35)	<i>p</i> -Coumaric acid derivative [56]	0.4 \pm 0.03	<0.001
4b ^a	5.71	272/ 319	325	MS ² [325]: 145 (100), 163 (7), 161 (2), 119 (1)	<i>p</i> -Coumaric acid glucoside [53–55]	0.4 \pm 0.03	<0.001
5a ^a	5.87	273/ 315	1125	MS ² [1125]: 463 (100), 301 (69), 639 (21), 625 (17), 801 (6), 477 (1)	Quercetin tetrahexoside-glucuronide	1.09 \pm 0.05	<0.001
5b ^a	5.87	273/ 315	1155	MS ² [1155]: 463 (100), 301 (67), 625 (21), 669 (17), 831 (4), 151 (1)	Quercetin derivative	1.09 \pm 0.05	<0.001
6	6.15	271/ 302	1139	MS ² [1139]: 815 (100), 653 (69), 301 (43), 463 (42), 977 (18), 695 (15), 625 (5)	Quercetin trihexoside-diglucuronide	1.16 \pm 0.06	<0.001
7	6.53	267/ 304	917	MS ² [917]: 300 (100), 301 (14), 917 (51), 179 (5), 771 (1)	Quercetin <i>p</i> -coumaroyl-rutinoside-hexoside [33–35]	7.2 \pm 0.2	<0.001
8	6.63	231/ 304	369	MS ² [369]: 189 (100), 207 (59), 145 (59), 369 (41), 163 (15), 119 (11)	<i>p</i> -Coumaric acid derivative [56]	2.59 \pm 0.06	<0.001
9a ^a	8.27	289	475	MS ² [475]: 167 (100), 135 (69), 125 (51), 443 (16), 247 (7), 143 (7)	Vanillic acid rutinoside [51,52]	5.71 \pm 0.01	<0.001
9b ^a	8.27	289	521	MS ² [521]: 167 (100), 125 (62), 443 (40), 135 (39), 247 (20), 143 (12)	Vanillic acid derivative	5.71 \pm 0.01	<0.001
10a ^a	8.59	265/ 352	755	MS ² [755]: 300 (100), 901 (75), 755 (19), 301 (10), 609 (1)	Quercetin rutinoside-rhamnoside [37–39]	8.2 \pm 0.5	<0.001
10b ^a	8.59	265/ 352	901	MS ² [901]: 284 (100), 285 (25), 737 (7), 179 (6), 755 (2), 593 (1)	Kaempferol dirutinoside [45, 46]	8.2 \pm 0.5	<0.001
11a ^a	11.94	264/ 356	463	MS ² [463]: 301 (100), 463 (36), 151 (10), 179 (5), 341 (5), 300 (6)	Quercetin hexoside [31,40–42]	2.7 \pm 0.1	<0.001
11b ^a	11.94	264/ 356	739	MS ² [739]: 284 (100), 739 (24), 285 (18), 575 (4), 179 (3), 593 (1)	Kaempferol rutinoside-rhamnoside [38,47]	2.7 \pm 0.1	<0.001
12a ^a	12.35	268/ 344	769	MS ² [639]: 314 (100), 315 (19), 769 (22), 299 (10)	Isorhamnetin rutinoside-rhamnoside [48,49]	1.06 \pm 0.03	<0.001
12b ^a	12.35	268/ 344	1099	MS ² [1099]: 463 (100), 625 (25), 301 (43), 567 (7), 300 (2)	Quercetin derivative	1.06 \pm 0.03	<0.001
13	13.19	259/ 351	609	MS ² [609]: 300 (100), 301 (54), 609 (29)	Quercetin rutinoside (Rutin) ^(Std)	0.78 \pm 0.01	<0.001
14a ^a	14.11	273/ 321	1361	MS ² [1361]: 301 (100), 463 (75), 669 (65), 625 (16), 831 (6), 300 (6), 1037 (2)	Quercetin derivative	1.4 \pm 0.1	<0.001
14b ^a	14.11	273/ 321	755	MS ² [755]: 285 (100), 755 (25), 284 (24), 593 (1)	Kaempferol rutinoside-hexoside ¹⁵⁰	1.4 \pm 0.1	<0.001
15	14.33	274/ 319	463	MS ² [463]: 300 (100), 301 (27), 463 (1)	Quercetin-3-O-glucoside ^(Std)	0.65 \pm 0.01	<0.001
16	14.83	300	303	MS ² [303]: 125 (100), 285 (45), 177 (16)	Taxifolin ^(Std)	1.00 \pm 0.02	<0.001
17	15.08	298	207	MS ² [207]: 119 (100), 207 (16), 163 (10)	<i>p</i> -Coumaric acid derivative	0.7 \pm 0.1	<0.001
18	22.95	264/ 314	901	MS ² [901]: 300 (100), 301 (12), 755 (45), 179 (4), 609 (1)	Quercetin dirhamnoside-rutinoside [43]	14.4 \pm 0.6	<0.001
19	25.59	315	885	MS ² [885]: 284 (100), 739 (69), 285 (21), 575 (6), 179 (3), 593 (2)	Kaempferol dirhamnoside-rutinoside [43,50]	8.0 \pm 0.4	<0.001
Total phenolic acids						9.4 \pm 0.1	<0.001
Total flavonoids						58.5 \pm 1	<0.001
Total phenolic compounds						68 \pm 1	<0.001

^a – Correspond to coeluted compounds in the peak of the same number; Std – Standard; Mean statistical differences obtained by *t*-Student test.

Da +146 Da) and 176 Da, ascribed as rhamnose, hexose, dihexoses, rutinoses and glucuronide moieties, respectively [31,33–43]. The presence of rutinose group was inferred, as molecules containing this sugar typically exhibit a less fragmented mass spectrum compared to those of neohesperidose [27,32,44]. Rhamnose was tentatively assigned as the deoxyhexose group in the molecules due to its abundance in nature [44]. Therefore, these peaks were assigned as quercetin trihexoside (**Peak 2**; λ_{\max} = 267/327 nm; [M – H][–] *m/z* 787), quercetin tetrahexoside-rhamnoside (**peak 3**; λ_{\max} = 271/312 nm; [M – H][–] *m/z* 1095), quercetin tetrahexoside-glucuronide (**peak 5a**; λ_{\max} = 273/315 nm; [M – H][–] *m/z* 1125), quercetin derivative (**peak 5b**; λ_{\max} = 273/315 nm; [M – H][–] *m/z* 1155), quercetin trihexoside-diglucuronide (**peak 6**; λ_{\max} = 271/302 nm; [M – H][–] *m/z* 1139), quercetin rutinoside-rhamnoside (**peak 10a**; λ_{\max} = 265/352 nm; [M – H][–] *m/z* 755), quercetin hexoside (**Peak 11a**; λ_{\max} = 264/356 nm; [M –

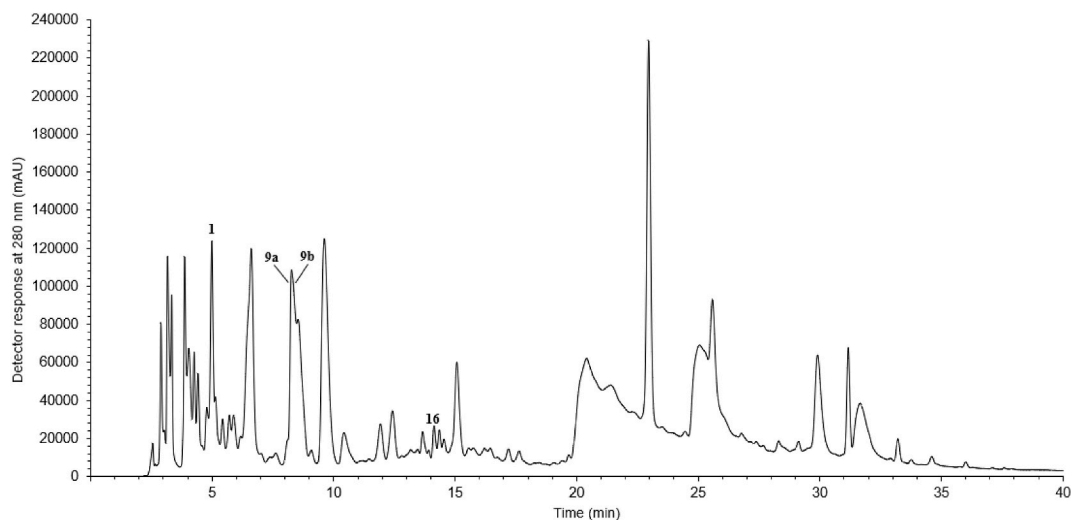


Fig. 1. Chromatogram of phenolic compounds present in the methanolic extract of *L. annua* seeds at 280 nm. Information regarding peak numbers and respective compounds is presented in Table 3.

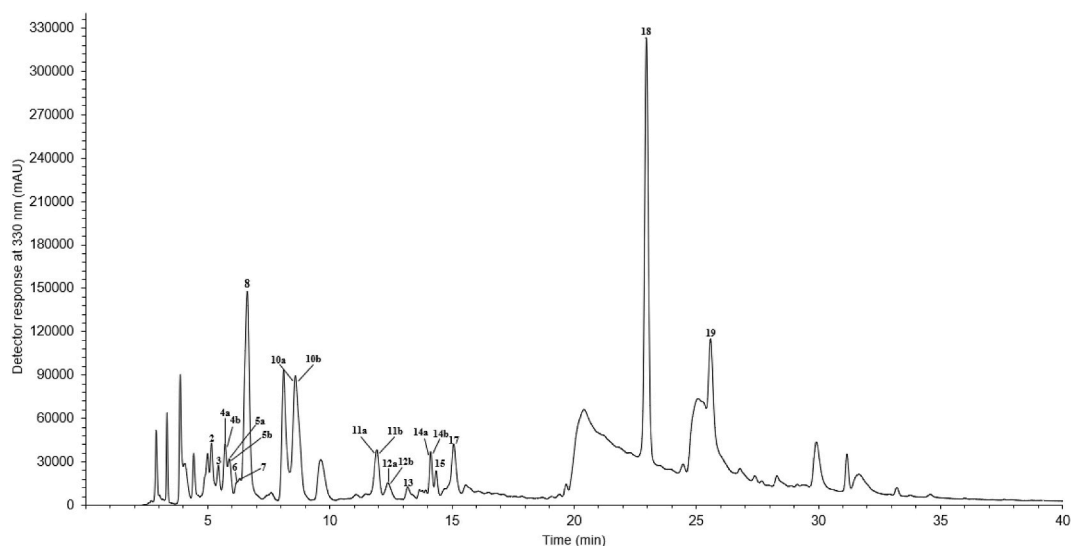


Fig. 2. Chromatogram of phenolic compounds present in the methanolic extract of *L. annua* seeds at 330 nm. Information regarding peak numbers and respective compounds is presented in Table 3.

H^- m/z 463), quercetin derivative (**peak 12b**; $\lambda_{\text{max}} = 268/344$ nm; $[\text{M} - \text{H}]^-$ m/z 1099), quercetin derivative (**peak 14a**; $\lambda_{\text{max}} = 273/321$ nm; $[\text{M} - \text{H}]^-$ m/z 1361) and quercetin dirhamnoside-rutinoside (**peak 18**; $\lambda_{\text{max}} = 264/314$ nm; $[\text{M} - \text{H}]^-$ m/z 901) [18, 24–31, 33, 34]. **Peak 7** ($\lambda_{\text{max}} = 267/304$ nm; $[\text{M} - \text{H}]^-$ m/z 917) gave deprotonated molecule at m/z 917 and fragment ions at m/z 300, 301 and 771, and was tentatively designated as quercetin *p*-coumaroyl-rutinoside-hexoside [33–35].

Four kaempferol derivatives were tentatively identified due to the presence of the aglycone ions at m/z 285 and 284, and the characteristic losses of sugar moieties (162 Da, 146 Da and 308 Da). The compounds were assigned as kaempferol dirutinoside (**peak 10b**; $\lambda_{\text{max}} = 265/352$ nm); deprotonated ion at m/z 901 and diagnostic fragments at m/z 593 and 285 $[\text{M} - \text{H} - 308 - 308]^-$, kaempferol rutinoside-rhamnoside (**peak 11b**; $\lambda_{\text{max}} = 264/356$ nm; with $[\text{M} - \text{H}]^-$ at m/z 739 and main fragment at m/z 285 $[\text{M} - \text{H} - 308 - 146]^-$), kaempferol rutinoside-hexoside (**peak 14b**; $\lambda_{\text{max}} = 273/321$ nm; molecular ion at m/z 755 giving ion fragment at m/z 285 $[\text{M} - \text{H} - 308 - 162]^-$), and kaempferol dirhamnoside-rutinoside (**peak 19**; $\lambda_{\text{max}} = 268/315$ nm) showing deprotonated molecule at m/z 885 and main fragment at m/z 284 and 285 $[\text{M} - \text{H} - 146 - 146 - 308]^-$ [38, 45–50].

Compound **12a** showed fragment molecular ion at m/z 769 and m/z 314, 315 and 299 as main fragment ions, representing the aglycone isorhamnetin after the loss of 454 Da due to the cleavage of a rutinoside (308 Da) and a rhamnoside (146 Da) [48, 49]. Hence, **peak 12a** was named as isorhamnetin rhamnoside-rutinoside ($\lambda_{\text{max}} = 268/344$ nm). Compounds **9a** ($\lambda_{\text{max}} = 289$ nm; $[\text{M} - \text{H}]^-$ m/z 475) and **9b** ($\lambda_{\text{max}} = 289$ nm; $[\text{M} - \text{H}]^-$ m/z 521) were tentatively identified as vanillic acid rutinoside and vanillic acid derivative,

Table 4
Antimicrobial activity of *L. annua* extracts (mg/mL).

Antibacterial activity		<i>S. aureus</i> (ATCC 11632)	<i>B. cereus</i> (clinical isolate)	<i>L.monocytogenes</i> (NCTC 7973)	<i>S.Typhimurium</i> (ATCC 13311)	<i>E. coli</i> (ATCC 5922)	<i>E. cloacae</i> (ATCC 35030)
Dichloromethane extract	MIC	1.00	1.00	1.00	1.00	0.50	1.00
	MBC	2.00	2.00	2.00	2.00	1.00	2.00
Methanol extract	MIC	1.00	1.00	1.00	1.00	0.25	1.00
	MBC	2.00	2.00	2.00	2.00	0.50	2.00
E211	MIC	4.00	0.50	1.00	1.00	1.00	2.00
	MBC	4.00	0.50	2.00	2.00	2.00	4.00
E224	MIC	1.00	2.00	0.50	1.00	0.50	0.50
	MBC	1.00	4.00	1.00	1.00	1.00	0.50
		<i>A.fumigatus</i> (ATCC 9197)	<i>A.versicolor</i> (ATCC 11730)	<i>A.ochraceus</i> (ATCC 6275)	<i>P.funiculosum</i> (ATCC 36839)	<i>P.verrucosum</i> var. <i>cyclopium</i> (food isolate)	<i>T.viride</i> (IAM 5061)
Dichloromethane extract	MIC	0.50	1.00	1.00	1.00	1.00	0.50
	MFC	1.00	2.00	2.00	2.00	2.00	1.00
Methanol extract	MIC	1.00	0.50	1.00	1.00	0.50	1.00
	MFC	2.00	1.00	2.00	2.00	1.00	2.00
E211	MIC	1.00	2.00	1.00	1.00	2.00	1.00
	MFC	2.00	4.00	2.00	2.00	4.00	2.00
E224	MIC	1.00	1.00	1.00	0.50	1.00	0.50
	MFC	1.00	1.00	1.00	0.50	1.00	0.50
		<i>C. albicans</i> 475/15	<i>C. albicans</i> 13/15	<i>C. albicans</i> 17/15	<i>C. parapsilosis</i> ATCC 22019	<i>C. tropicalis</i> ATCC 750	<i>C. krusei</i> H1/16
Dichloromethane extract	MIC	0.50	1.00	0.50	1.00	1.00	0.50
	MFC	1.00	2.00	1.00	2.00	2.00	1.00
Methanol extract	MIC	1.00	0.50	1.00	1.00	0.50	1.00
	MFC	2.00	1.00	2.00	2.00	1.00	2.00
Ketoconazole (x 10⁻³)	MIC	3.20	1.60	1.60	3.20	1.60	1.60
	MFC	6.40	51.20	51.20	6.40	6.40	3.20

Table 5
Antioxidant activity of *L. annua* seeds oil extracts (mean ± SD).

	TBARS (IC ₅₀ , mg/mL)	Student <i>t</i> -Test
Dichloromethane	1.25 ± 0.02	<0.001
Methanol	0.32 ± 0.01	<0.001
Trolox	0.0054 ± 0.0003	<0.001

Mean statistical differences obtained by *t*-Student test.

respectively, due to the presence of the main fragment at *m/z* 167 (vanillic acid), indicating the loss of a rutinoside moiety (308 Da), in addition to other characteristic fragments such as *m/z* 125, 443, and 135 [51,52]. Derivatives of *p*-coumaric acid were also observed, showing typical aglycone ion at *m/z* 163, and characteristic fragments such as at *m/z* 145 and 119, being assigned as *p*-coumaric acid derivative (**peak 4a**; λ_{\max} = 272/319 nm; [M – H][–] *m/z* 369), *p*-coumaric acid glucoside (**peak 4b**; λ_{\max} = 272/319 nm; [M – H][–] *m/z* 325), *p*-coumaric acid derivative (**peak 8**; λ_{\max} = 231/304 nm; [M – H][–] *m/z* 369), *p*-coumaric acid derivative (**peak 17**; λ_{\max} = 298/369 nm; [M – H][–] *m/z* 207) [53–56].

Both tested extracts displayed good antimicrobial activity towards tested microorganisms (Table 4). Among bacteria, *E. coli* showed the highest susceptibility to dichloromethane and methanolic extracts activity (MIC 0.25 mg/mL, MBC 0.50 mg/mL), whereas rest of the tested bacteria showed quite uniform susceptibility to the tested extracts. In comparison with positive controls E211 and E244, extracts displayed comparable or better activity which may be of importance in the food technology sector. The observed pattern of antibacterial values also refers to our antifungal results, since MIC value for fungi ranged from 0.50 to 1.00 mg/mL, whereas range of MFC values was in 1.00–2.00 mg/mL. Tested *Candida* spp. showed approximately the same sensitivity to the tested samples.

Strong antimicrobial potential observed in tested samples can be attributed to the presence of both phenolic compounds and fatty acids in our extracts, given that Khadke et al. [57] previously showed that fatty acids may be responsible for this type of activity. Thus, erucic acid showed antibacterial activity towards some *Borrelia* species [58], while oleic and nervonic acid (all abundantly present in seeds of *L. annua*) showed effects on biofilm formation in MDR *Acinetobacter baumannii*. Along with fatty acids, various phenolic compounds (identified in our extract) were associated with antimicrobial activity as well, i.e. flavonoids emerged as effective antimicrobial agents, with structural modifications influencing their efficacy [59]. Furthermore, synergistic interactions of phenolic compounds with antibiotics enhanced their overall antimicrobial potential, highlighting their potential in development of effective therapeutic strategies towards antibiotic-resistant pathogens [60]. Future research should include investigation of underlying mechanisms of action, as to choose wisely the most suitable candidates for development of antimicrobial drugs [61].

Results of TBARS assay are presented in Table 5. They demonstrate better antioxidant activity for methanol extract than

Table 6
Cytotoxic activity of *L. annua* seeds oil extracts (mean \pm SD).

<i>L. annua</i> extracts	IC ₅₀ value
Dichloromethane	>400 μ g/mL
Methanol	>400 μ g/mL
K ₂ Cr ₂ O ₇	83.40 μ g/mL

dichloromethane: EC₅₀ = 0.32 mg/mL and 1.25 mg/mL, in that order. Even so, results were \sim 100 fold lower than we obtained for the positive control Trolox (0.0054 mg/mL). Previously published data suggested that aerial parts and roots of *L. annua* have antioxidant properties as well. For aerial parts total antioxidant activity was 215.56 mg AAE/g, DPPH activity was 578.9 μ g/mL, ABTS⁺ was 648.3 μ g/mL, while inhibition of lipid peroxidation was 558.3 μ g/mL [2]. Roots of *L. annua* showed app. twice the value of aerial parts: total antioxidant activity 110.07 μ g/mL, DPPH activity was 1294.1 μ g/mL, ABTS⁺ was 1072.5 μ g/mL, and inhibition of lipid peroxidation 342.7 μ g/mL [2]. We tested lipid peroxidation using TBARS assay and found better activity in case of methanolic extract (0.32 mg/mL) than the one demonstrated by Katanić Stanković et al. (2022) [2]. Better results obtained with methanolic extract than with dichloromethane can be explained with higher polarity of methanol, and consequently extraction of phenolic compounds responsible for the antioxidant potential, which we also demonstrated by analysing the phenolic compounds.

Cytotoxic activity results of *L. annua* extracts are presented in Table 6. They suggest that the tested extracts have no cytotoxic effect on spontaneously immortalized HaCaT cell line (IC₅₀ > 400 μ g/mL).

Based on results regarding bioactive potential of *L. annua*, we can conclude that this plant exhibits great versatility and may have applications in both food and pharmaceutical industries, providing natural solutions for a variety of purposes. Thus, results we obtained in antioxidant assay regarding lipid peroxidation as well as those demonstrating antimicrobial potential towards food borne pathogens, suggest that *L. annua* has the potential to enhance safety of food ingredients, address public concerns related to spoilage and rancidity, and decrease use of synthetic preservatives [62].

Moreover, given the results of antibacterial activity observed towards human pathogens that cause skin infections, we further showed that *L. annua* has no toxic effects on HaCaT cells, suggesting potential application of *L. annua* extracts in development of skincare products.

4. Conclusion

This study investigated the untapped potential of *L. annua*, an understudied yet versatile species with a range of applications in agriculture, ecology and medicine. Despite its prominence as an ornamental plant *L. annua*, could offer significant benefits as a sustainable crop, contributing to agricultural diversification, soil health, and pollinator support. Its edible parts and potential medicinal properties further enhance its appeal. This research uncovered new insights into its versatile applications, while also highlighting its connection to sustainable farming practices, ultimately aiming to promote its integration into modern agricultural systems. The results suggest *L. annua* has a favorable nutritional/chemical profile, with significant antioxidant and antibacterial potential while displaying no cytotoxic effects. These features highlight its potential to be regarded as functional food and expand its possible applications in both food and pharmaceutical industries. Furthermore, given the low cultivation requirements and possibility to grow it on abandoned farmlands, our primary goal is to establish *Lunaria annua* as a cultivated crop which will allow extensive use of this underutilized plant, ensuring a sustainable and consistent supply of its bioactive compounds, thus facilitating the cosmopolitan use of this versatile plant.

CRedit authorship contribution statement

Jovana Petrović: Writing – original draft, Investigation, Formal analysis, Conceptualization. **Daiana Almeida:** Writing – original draft, Investigation, Formal analysis, Data curation. **Ángela Fernandes:** Investigation, Formal analysis, Data curation. **Dejan Stojković:** Writing – review & editing, Investigation, Conceptualization. **Dragana Robajac:** Writing – review & editing, Formal analysis, Data curation. **Tayse F.F. da Silveira:** Methodology, Formal analysis. **Lillian Barros:** Writing – review & editing, Supervision, Project administration, Funding acquisition, Conceptualization.

Data availability statement

The data that support the findings of this study are available on request from the corresponding author, [J.P.].

Declaration of competing interest

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: Corresponding author Dejan Stojković serves as Associate Editor of the journal Heliyon. If there are other authors, they declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix ASupplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.heliyon.2025.e42248>.

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